



Assessing Chemical Variations in *Andrographis paniculata*: GC-MS Characterization of Its Hexane Extract

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Abstract

Phytochemical analysis and characterization are essential for identifying potential therapeutic compounds in plants, enabling drug discovery, validating traditional medicine, and ensuring the therapeutic value and reliability of herbal products. The results revealed the presence of various bioactive compounds, including saponins, flavonoids, terpenoids, and alkaloids, in the hexane extract, aligning with previous research. Notably, tannins, steroids, and cardiac glycosides were not present. GC-MS characterization identified 19 distinct compounds in the hexane extract, each with unique retention times and abundance levels. Squalene, a terpenoid, was the most abundant component (39.927%), followed by 3, 7, 11, 15-tetranethy-2-hexadecent-1-ol (14.208%) and Stigmastan-3-ol derivatives. Interestingly, the extract displayed a diverse range of chemical classes, with nine major alkaloids, six major terpenoids, and two major waxes identified. Among the alkaloids, one of the isomers of Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl- exhibited the highest abundance (4.973%).

Keywords: *Andrographis Paniculata*, Hexane Extract, GC-MS Analysis, Phytochemical Composition, Bioactive Compounds

Introduction

Chemical analysis of plant extracts indicates possible bioactivities for further investigation and enables the identification of chemical markers. Because they prevent degenerative illnesses, flavonoids, for instance, are secondary metabolites with significant therapeutic significance. Cytotoxic, thrombolytic, anti-inflammatory, anticancer, vasorelaxant, and antioxidant actions are among the biological characteristics of these substances that have been reported (Islam et al., 2013; Cabral et al., 2016). Considered significant antioxidant agents are also other phenolic compounds like tannins and anthocyanins. Because they shield the body against oxidative stress and related illnesses, natural antioxidants are crucial to medical treatment. Additionally regarded as a significant source of bioactive chemicals against viral illnesses are natural products (Siqueira et al., 2016). Over the past 30 years, there has been a gradual increase in fungal infections, making them one of the most urgent problems with public health. (da Costa Cordeiro et al., 2018; Park et al., 2020; Samuel et al., 2023).

In the vibrant tapestry of natural remedies and herbal medicine, there exists a treasure trove of botanical wonders, each with its own unique story to tell. Among these, *Andrographis paniculata*, hailed as the 'King of Bitters,' stands as a testament to the healing potential that Mother Nature has to offer. This unassuming herbaceous plant, with its lanceolate green leaves and a taste as bitter as its reputation has captivated the attention of herbalists and scientists alike for generations (Okhwarobo et al., 2014; Ushie et al., 2019; Samuel et al., 2023). But what secrets lie within the leaves of *Andrographis paniculata*? What chemical symphony orchestrates its remarkable medicinal properties? To unravel these mysteries, this research provides insight into the world of phytochemicals and bioactive compounds hidden within the very essence of this botanical marvel (Andrew et al., 2018; Andrew et al., 2017).

Andrographis paniculata, a member of the Acantheceae family, is a small shrub genus with 28 known species. This annual herbaceous plant, known for its extreme bitterness, features lanceolate green leaves, white flowers with rose-purple spots on the petals, and typically grows to a height of 60-70cm. It goes by various names worldwide, including "Mahatikta" in North Eastern India, "Chirota" in Arabic, "Chuanxinlian" in Chinese, "Nelaveva" in Indonesia, "Kalmegh" or "Kalamegha" in India, "Hempedu Buni" in Malaysia, "Vinegar" in Akwa-Ibom Nigeria, and "Nila-vembu" in Tamil. Found abundantly in Asian countries like India, Sri Lanka, Pakistan,

Java, Malaysia, China, Hong Kong, the Philippines, Thailand, and Indonesia, this plant has a significant presence in traditional medicine (Ullah & Hassan, 2021; Asuqwo & Etim, 2011).

Phytochemicals, naturally occurring compounds in plants, offer health benefits such as antioxidants, anti-inflammatory, and anti-cancer properties (Srinivasan et al., 2007). *Andrographis paniculata*, renowned for its medicinal properties, particularly in its leaves and stems, exhibits antimicrobial, antioxidant, antiallergic, anti-inflammatory, and anti-cancer effects (Misira et al., 1990; Zhang and Tan, 1997; Calabrese et al., 2010; Gupta et al., 2012; Deepak et al., 2014). Initial screening identified flavonoids, alkaloids, steroids, phenols, and tannins in the leaves (Deepak et al., 2014), and another study by Kalaivani et al. (2012) found alkaloids, steroids, tannins, and saponins in chloroform, petroleum ether, and ethanol extracts. However, the specific phytochemicals responsible for these properties remain largely unidentified. This research aims to systematically characterize the phytochemicals in *Andrographis paniculata* leaves (Samuel et al., 2024; Ushie et al., 2013).

The analytical technique known as Gas Chromatography-Mass Spectrometry (GC-MS) blends the material identification properties of mass spectrometry with the separation powers of gas chromatography (Garcia, & Barbas, 2011; Inyang et al., 2018). The compounds are separated based on interactions with the column's coating and carrier gas, and then converted into ions. For solid samples, extraction, outgassing, or pyrolysis is used. GC-MS applications span drug detection, environmental analysis, explosives detection, and identifying unknown substances. It has been employed in various fields, such as airport security and Mars probe missions, due to its sensitivity. Recent studies on *Andrographis paniculata* extracts using GC-MS identified bioactive compounds with potential pharmaceutical and industrial applications, showcasing its significance in research and industry (Chauhan et al., 2014; Asuqwo et al., 2011; Sivakumar & Gayathri, 2015). Recent research on *Andrographis paniculata* reveals the presence of key bioactive compounds, including 14-deoxy-11-oxoandrographolide, andrographolide D, 14-deoxyandrographolide, neoandrographolide, andrographosterol, and others, with andrographolide being responsible for the plant's bitter taste. Andrographolide is in all parts of the plant (Deshpande et al., 2014; Asuqwo et al., 2012). This study conducts a phytochemical screening of *Andrographis paniculata* leaves and seeks to identify and quantify the specific chemical components of the phytochemicals within hexane extracts using the advanced analytical capabilities of GC-MS (Etim et al., 2017; . The results of this study hold substantial significance as they contribute to the field of phytochemistry and pharmacology. It would provide a much-needed comprehensive overview of the phytochemicals and bioactive compounds in hexane-extracted leaves of *A. paniculata*. This information could be used to develop new therapeutic agents from this plant and improve the understanding of its medicinal properties.

Materials and Methods

Sample Collection: *Andrographis paniculata* was collected from Kom-Kom in Oyigbo L.G.A of Rivers State in the morning and promptly sealed in a polyethylene bag. For identification, it was then sent to the Department of Plant Science and Biotechnology at the University of Port-Harcourt.

Sample Preparation: *Andrographis paniculata* leaves were removed from branches, washed to remove dust, and air-dried at room temperature for two weeks. Then, in order to maximize the sample's extraction surface area, they were mixed into a homogenous semi-powder.

Extraction Procedure: The process used to obtain the hexane extracts involved macerating about 35g of semi-powdered *Andrographis paniculata* in hexane for 48 hours while shaking it occasionally. After the material was soaked and filtered, the procedure was carried out again using water, ethyl acetate, and hexane in that order.

Phytochemical Analysis

Tannins Test: In separate test tubes, 4 drops of FeCl_3 were added after 1 mL of hexane extract and 5 mL of distilled water were combined. The mixture was watched using ferric salts to develop a deep brown or bluish-black hue (Finar, 2008; Etim et al., 2022).

Saponins (Frothing Test): 1 ml of hexane extract was combined with 1 ml of distilled water in test tubes. The mixture was vigorously shaken for 2 minutes and observed for continuous foaming.

Flavonoids Test: In a test tube with 5 ml of diluted NaOH, 1 ml of hexane extract was dissolved. Excessive drops of diluted HCl were then added. The mixture was observed for the development of an orange, yellow, or violet coloration (Finar, 2008; Asuqwo et al., 2010).

Glycosides (Lieberman's Test): After dissolving 0.5 ml of hexane extract in 2 ml of acetic anhydride in separate test tubes, the mixture was allowed to cool. Subsequently, droplets of conc H_2SO_4 were added, and the mixture's red hue was noticed before it turned green.

Steroids (Lieberman's Test): In test tubes, 0.5 ml of hexane extract and 2 ml of acetic anhydride were mixed, and then the mixture was allowed to cool. After adding drops of concentrated H₂SO₄, the mixture was checked for a greenish tint (Oko et al., 2017; Onen et al., 2017; Nweke-Maraizu et al., 2023).

Terpenoids (Lieberman's Test): After dissolving 0.5 ml of hexane extract in 2 ml of acetic anhydride in separate test tubes, the mixture was allowed to cool. Drops of concentrated H₂SO₄ were added, and the mixture's brown color was noted.

Alkaloids Test: After treating 1 mL of hexane extract in test tubes with 2 mL of Wagner's reagent, the production of an insoluble precipitate was monitored (Finar, 2008; Khan et al; 2021; Khan et al., 2023).

GC-MS Analysis

An Agilent 7890A-5975C GC-MS system with an HP5-column was used to investigate the hexane extracts of *Andrographis paniculata*. With the injector and ion source temperatures set at 250°C and 280°C, respectively, an electron ionization system operating at 70eV was employed. From 110°C to 200°C at 10°C/min, and then to 280°C at 5°C/min, the oven temperature was programmed and maintained for nine minutes. At 70eV, mass spectra were collected. Peak area was used to determine the percent composition.

Identification of Components

Based on GC retention time, organic components in the extracts were identified and cross-referenced with the NIST Library. In order to make qualitative conclusions, peak regions were compared to total ion current (TIC) areas while supplying details on the names, molecular weights, retention periods, percentages, and structures of the compounds.

Results

Analysis of Selected Phytochemicals of Leaves of *Andrographis paniculata* Hexane Extracts

Table 1: Phytochemical Analysis of Leaves of *Andrographis paniculata* Hexane Extracts.

Phytochemicals	Hexane Extract
Tannins	-
Saponins	+
Flavonoids	+
Steroids	-
Terpenoids	+
Alkaloid	+
Cardiac Glycoside	-

Key: (+) Present (-) Absent

Table 2. Chemical Composition of Some Phytochemicals Found in Hexane Extract of *Andrographis paniculata*.

Phytochemical	S/N	Names of Compound	Molecular Formula	MW	%	RT
Alkaloid	1	I-Guanidinosuccinimide	C ₅ H ₇ N ₃ O ₂	141	1.688	2.627
	2	Benzeneethanamine, 2, 5-difluoro-β, 3, 4-trihydroxy-N-methyl-	C ₉ H ₁₁ F ₂ NO ₃	219	4.973	3.905
	3	Benzeneethanamine, 2, 5-difluoro-β, 3, 4-trihydroxy-N-methyl-	C ₉ H ₁₁ F ₂ NO ₃	219	1.364	4.510
	4	Benzeneethanamine, 2, 5-difluoro-β, 3, 4-trihydroxy-N-methyl-	C ₉ H ₁₁ F ₂ NO ₃	219	2.147	4.761
	5	Benzeneethanamine, 2, 5-difluoro-β, 3, 4-trihydroxy-N-methyl-	C ₉ H ₁₁ F ₂ NO ₃	219	3.688	9.842

	6	Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl-	$C_9H_{11}F_2NO_3$	219	1.344	10.950
	7	Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl-	$C_9H_{11}F_2NO_3$	219	1.405	13.148
	8	Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl-	$C_9H_{11}F_2NO_3$	219	1.892	15.073
	9	Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl-	$C_9H_{11}F_2NO_3$	219	1.991	16.614
	10	Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl-	$C_9H_{11}F_2NO_3$	219	2.130	18.149
Terpenoids	1	1, 2-Benzene dicarboxylic acid, bis (2-methyl propyl) ester	$C_{16}H_{38}O_2$	278	3.325	9.169
	2	11, 14-Eicosadienoic acid, methyl ester.	$C_{21}H_{38}O_2$	322	1.761	9.270
	3	α -Tocopheryl acetate	$C_{31}H_{52}O_2$	472	1.321	16.865
	4	Stigmastan-3- β , 5-chloro-, acetate, (3 β , 5 α)-	$C_{31}H_{52}ClO_2$	492	7.616	18.919
	5	3, 7, 11, 15-Tetramethyl-2-hexadecen-1-ol	$C_{20}H_{40}O$	296	14.208	10.762
	6	Squalene	$C_{30}H_{50}$	410	39.927	14.565
Flavonoids	1	2(IH)-Benzocyclooctenone, decahydro-10a-methyl-trans-	$C_{13}H_{22}O$	194	3.688	8.976
Wax	1	Tetradecane, 2, 6, 10-trimethyl-	$C_{17}H_{36}$	240	3.713	12.977
	2	Hentriacontane	$C_{31}H_{64}$	436	2.185	13

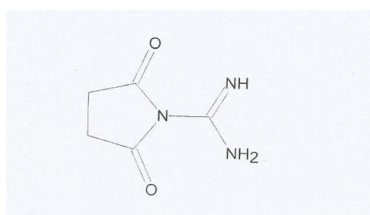
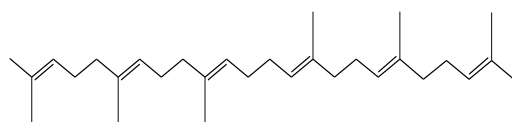
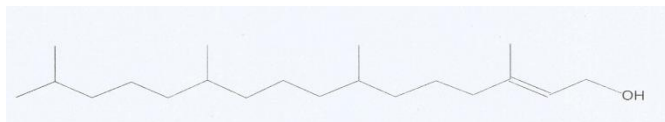


Figure 1. Structure of 1-Guanidinosuccinimide

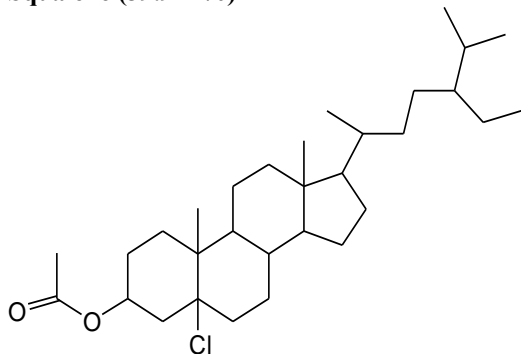
Prevalent Compound Structures in Hexane Extract



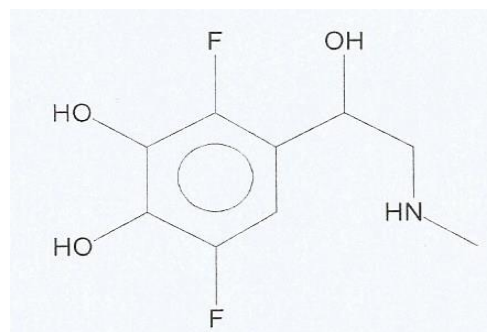
Squalene (39.927%)



3,7,11,15-tetramethyl-2-hexadecen-1-ol (14.208%)



Stigmastan-3-ol, 5-chloro- acetate, (3 β , 5 α)- (7.616%)



Benzeneethanamine, 2,5-Difluoro- β , 3,4-Trihydroxy-N-Methyl (4.973%)

Discussion

In the hexane extracts of *Andrographis paniculata* leaves as indicated in Table 1, saponins, flavonoids, terpenoids, and alkaloids were detected, while tannins, steroids, and cardiac glycosides were absent. *Andrographis paniculata* leaf extracts contain flavonoids, saponins, and tannins in accordance with earlier research by Sofowara et al. (1993), Deepak et al. (2014), and Kalavaini et al. (2012). These phytochemicals are associated with the plant's various biological activities, including antimicrobial (Gadisa & Tadesse, 2021), anti-diabetic (Askari et al., 2021), and anti-inflammatory properties, highlighting its therapeutic potential (Izuagbe, et al., 2022; Askari et al., 2021; Shinggu et al., 2023). In the GC-MS analysis of hexane extracts of *Andrographis paniculata* leaves (Table 2), 19 different compounds were detected. Each compound exhibited distinct retention times and abundance levels. Retention times, determined by compound polarity and boiling points, ranged from 2 to 22 minutes. Notably, the hexane extract was rich in compounds, including Squalene (39.927%), 3, 7, 11, 15-tetranethy-2-hexadecent-1-ol (14.208%), Stigmastan-3-ol, 5-chloro-, acetate, (3 β , 5 α)- (7.616%), and one of the isomers of Benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-Methyl- (4.973%). The hexane extract contained nine major alkaloids, six major terpenoids, and two major waxes, with the most abundant alkaloid being one of the isomers of benzeneethanamine, 2, 5-difluoro- β , 3, 4-trihydroxy-N-methyl- (4.973%), and the dominant terpenoid being squalene (39.927%).

Conclusion

This study utilized GC-MS analysis to examine the hexane extract of *Andrographis paniculata* leaves and determine its phytochemical composition. The analysis revealed the presence of saponins, flavonoids, terpenoids, and alkaloids, as indicated in Table 1, while tannins, steroids, and cardiac glycosides were not detected. Among the 19 compounds identified, Squalene stood out as the major constituent. These results offer important insights on potential therapeutic and pharmaceutical applications of *Andrographis paniculata*. Further investigation in this field may lead to the development of novel therapeutic agents and advancements in natural product chemistry and drug discovery.

Recommendations

1. Future research should consider expanding the analysis to include other solvents (e.g., methanol, ethanol) to capture a wider range of compounds. And correlate the identified compounds with known biological activities to assess the potential therapeutic applications of the extract.
2. Additionally, the chemical composition of *Andrographis paniculata* from different geographical regions can be compared to identify variations and potential factors influencing the content of bioactive compounds.

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